

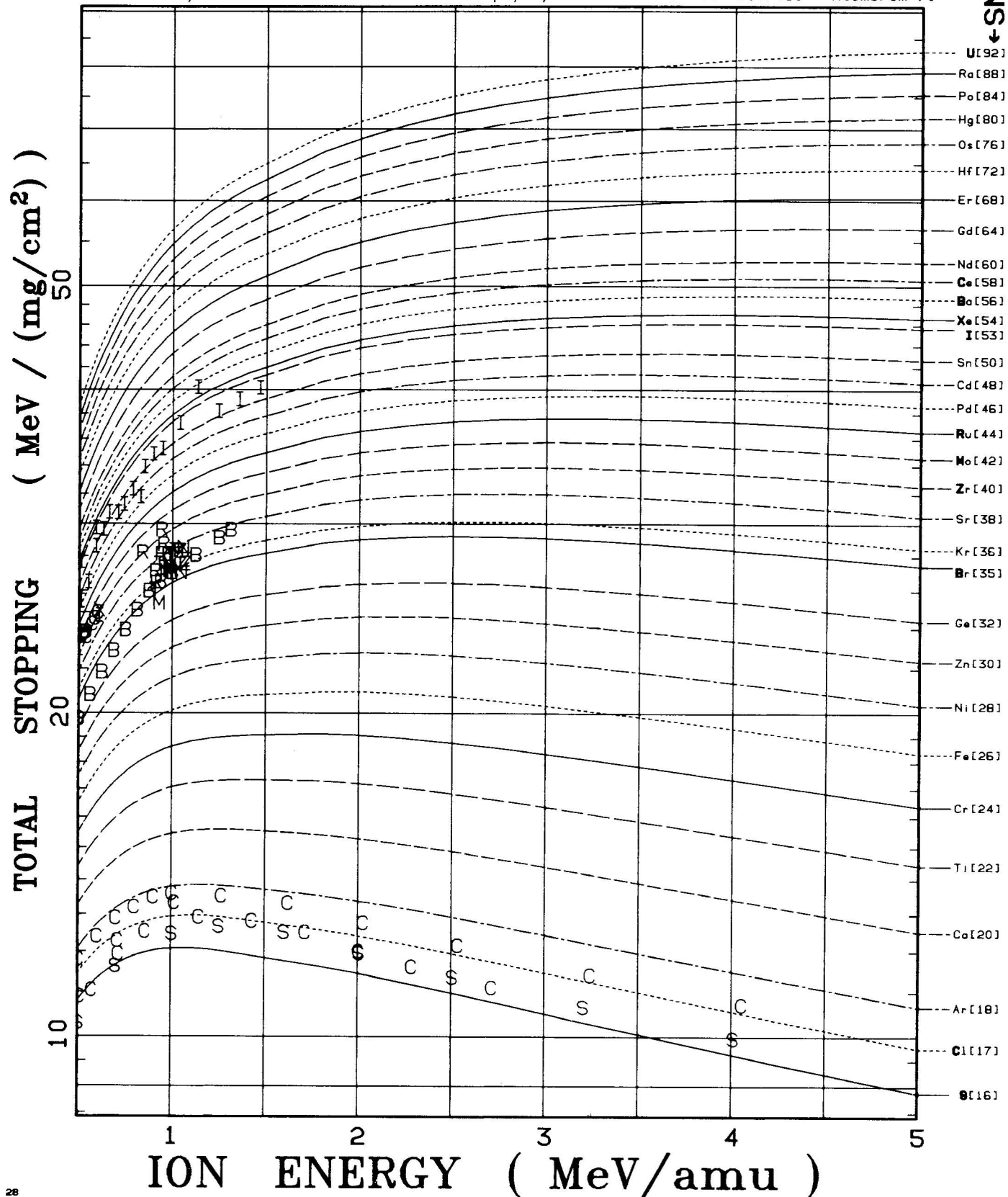
Ni(28)

←←← TARGET →→→

Ni(28)

Atom Density = 9.126×10^{22} Atoms/cm³ Multiply Total Stopping by 889.5 for Units: [MeV/mm]
 Mass Density = 8.895 Grams/cm³ Multiply by 97.49 for Units: [eV/(10¹⁵ Atoms/cm²)]

←SN01



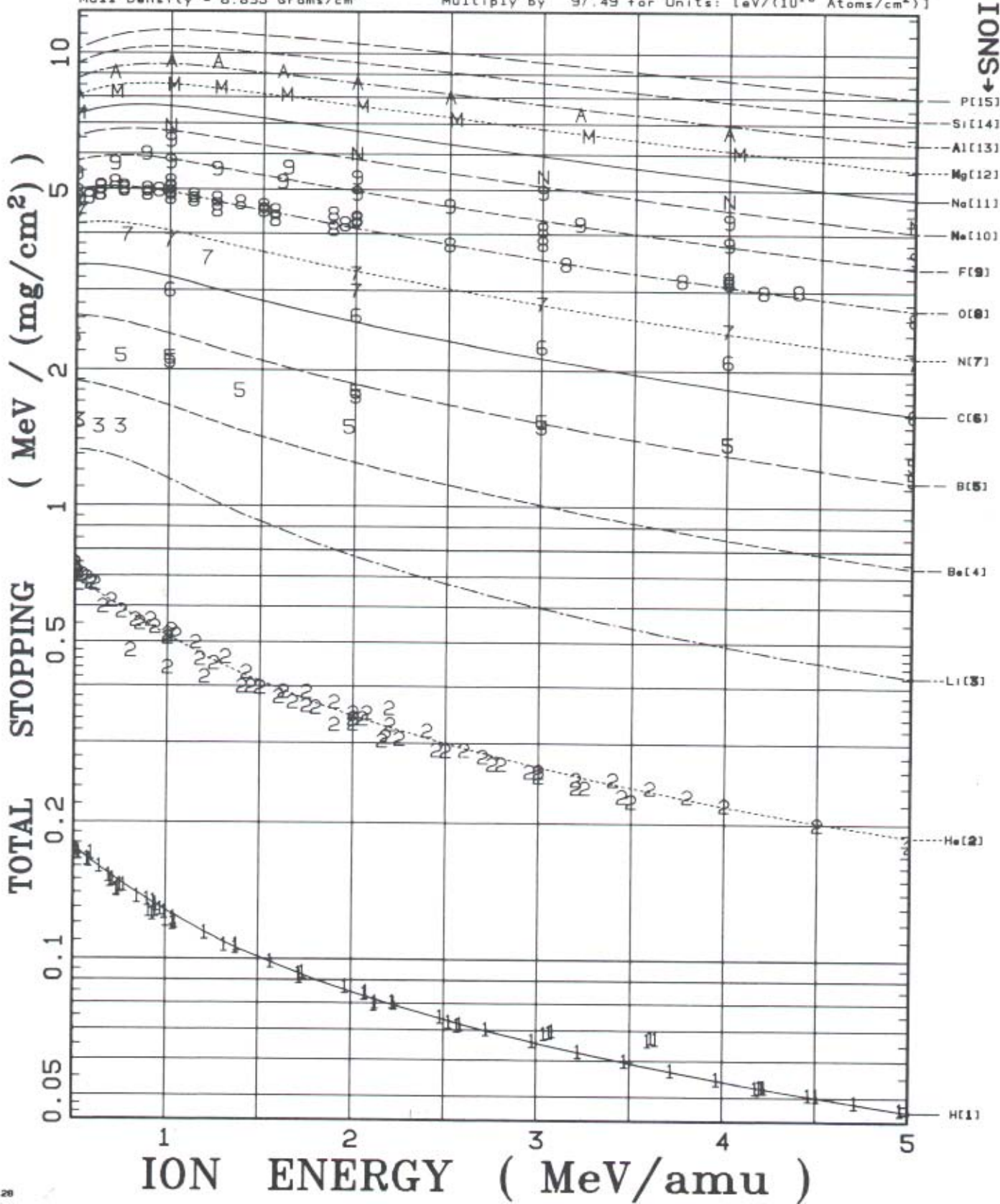
Ni(28)

←←← TARGET →→→

Ni(28)

Atom Density = 9.126×10^{22} Atoms/cm³
Mass Density = 8.895 Grams/cm³

Multiply Total Stopping by 889.5 for Units: [MeV/mm]
Multiply by 97.49 for Units: [eV/(10¹⁵ Atoms/cm²)]



Pb(82)

←←←

TARGET →→→

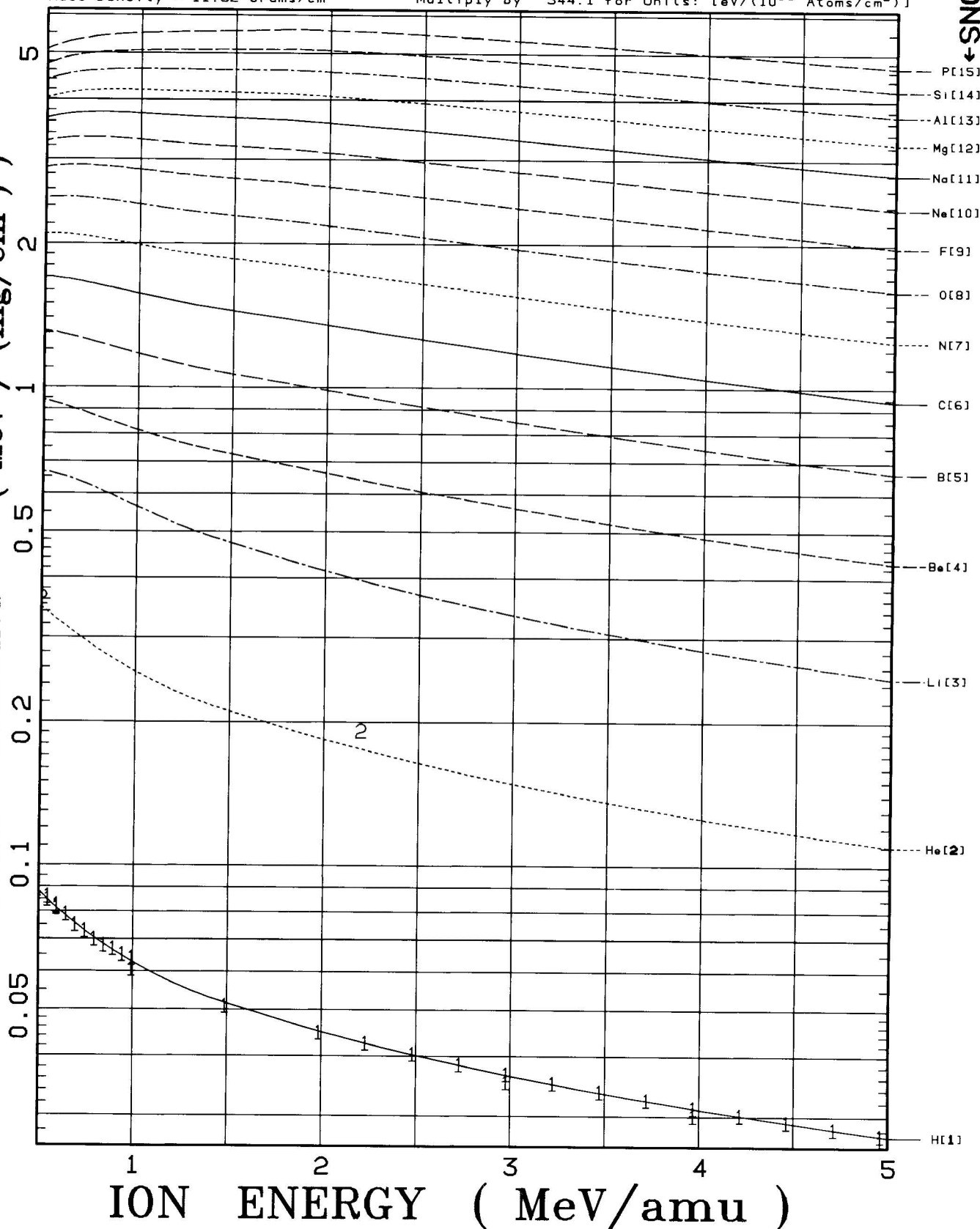
Pb(82)

Atom Density = 3.291×10^{22} Atoms/cm³Mass Density = 11.32 Grams/cm³

Multiply Total Stopping by 1132 for Units: [MeV/mm]

Multiply by 344.1 for Units: [eV/(10¹⁵ Atoms/cm²)]TOTAL STOPPING
(MeV / (mg/cm²))

←←← IONS



Pb(82)

←←←

TARGET

→→→

Pb(82)

IONS
↓SNOI

Atom Density = 3.291×10^{22} Atoms/cm³
Mass Density = 11.32 Grams/cm³

Multiply Total Stopping by 1132 for Units: [MeV/mm]
Multiply by 344.1 for Units: [eV/(10¹⁵ Atoms/cm²)]

